Supporting Information for

Stereocontrolled Total Synthesis of (±)-Catharanthine via Radical-Mediated Indole Formation

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General:

All reactions were carried out in oven-dried glassware under a slight positive pressure of argon unless otherwise stated. Reagents were either commercially available and used as obtained, or were prepared as noted or according to published methods. Benzene, toluene, and dichloromethane were distilled from CaH₂ and stored over activated molecular sieves (4Å). THF, ether, and acetonitrile, methanol, and ethanol were purchased anhydrous and stored over molecular sieves (4Å) under argon. t-Butanol and 1-propanol were 99%+ reagent grade and used as received. 'Workup', unless' otherwise noted, refers to partitioning the reaction mixture between the indicated aqueous and organic phases, followed by separation and extraction of the aqueous phase and washing of the combined organic phases as indicated. Organic solutions were dried with powdered anhydrous magnesium sulfate. Preparative flash column chromatography was performed using a quantity of silica gel (Merck Silica gel 60, 230-400 mesh) equal to 30 to 50 times sample weight, with gradient elution over the indicated range of solvent mixtures. Preparative thin layer chromatography (pTLC) was carried out on 200 x 65 x 0.5 mm precoated glass plates (Merck Silica gel 60 F₂₅₄). Yields, unless otherwise stated, refer to isolated yields of compounds judged greater than 95% pure by ¹H NMR. Previously reported compounds were identified on the basis of their ¹H NMR spectra, while new compounds were further characterized by ¹³C NMR and IR, and gave satisfactory high

resolution mass spectrographic analyses, unless noted otherwise. NMR spectra were obtained in CDCl3 on a JEOL LA-400 400 MHz spectrometer. All ¹H NMR spectra are reported in ppm downfield from tetramethylsilane as an internal standard. All ¹³C NMR spectra are reported in ppm relative to the central line of the triplet for CDCl₃ at 77 ppm. IR spectra were recorded on a JASCO FT/IR-410, and absorptions are reported in cm⁻¹. High resolution mass spectra were obtained on a JEOL JMS-GCmate MS-DIP20 quadrupole at 70 eV, using direct probe insertion at temperatures of 70 to 330 °C.

Experimental.

1-Benzyloxycarbonyl-3-ethyl-3,4-dehydropiperidine, 7.1 3-

Ethylpyridine (17.1 mL, 150 mmol) was cooled to 0 °C in an ice bath. Benzyl bromide (17.8 mL, 150 mmol) was added dropwise with stirring over several minutes. The reaction mixture evolved heat and turned yellow; after ca. 20 min, the reaction mixture grew viscous and rapidly solidified to a yellow glass. This was allowed to warm to room temperature and was left to stand overnight. The resulting opaque yellow-white solid was broken up with a spatula and ground in a mortar to produce a coarse, off-white powder. This was washed with several portions of ether and dried under vacuum to afford 1-benzyl-3-ethylpyridinium bromide as an off-white solid (41.58 g, 99.6%).

The pyridinium salt (6.00 g, 21.6 mmol) was dissolved in ethanol (35 mL) in an addition funnel, and added dropwise over 15 minutes to a stirred suspension of NaBH₄ (3.22g, 86.3 mmol) in ethanol (60 mL) cooled to 0 °C on

¹ Szántay, C.; Bölcskei, H.; Gács-Baitz, E. Tetrahedron 1990, 46, 1711.

an ice bath. Hydrogen evolved steadily durring the addition. The resulting bright-yellow mixture was allowed to warm slowly to room temperature overnight with stirring. The solvent was then removed under vacuum and the residue was worked up between water and dichloromethane (3x50 mL, brine) and dried. Concentration by rotary evaporation afforded crude 1-benzyl-3,4-dehydro-3-ethylpiperidine as an orange oil, 4.58 g.

The crude dehydropiperidine was dissolved in benzene. Benzyl chloroformate (6.17 mL, 43.2 mmol) was added in one portion via syringe. A slight exotherm was noted. The reaction mixture was heated at reflux for 4 h, at which time TLC analysis indicated complete conversion of the starting material. The solvent was removed under vacuum and the residue was purified by flash column chromatography (hexanes/ethyl acetate 15:1-5:1) to afford 1-benzyloxycarbonyl-3,4-dehydro-3-ethylpiperidine, 7, as a colorless oil (3.28 g, 62% from the pyridinium salt). ¹H NMR: 1.03 (t, *J*=7.3 Hz, 3H); 1.92 (br s, 2H); 2.13 (br s, 2 H); 3.52 (t, *J*=5.6 Hz, 2H); 3.86 (br s, 2H); 5.22 (s, 2H); 5.52 (br s, 1H); 7.35 (m, 5H).

1-Benzyloxycarbonyl-3,4-*trans*-dibromo-3-ethylpiperidine, **8.**¹ The Cbz-protected dehydropiperidine **7** (3.28 g, 13.4 mmol) was dissolved in reagent-grade dichloromethane (50 mL). Bromine (ca. 0.69 mL, 13.4 mmol) was added dropwise until the red color persisted. The mixture was stirred an additional 5 min., then decolorized with aqueous Na₂S₂O₃. The mixture was worked up with brine (1x20ml CH₂Cl₂) and dried. Concentration afforded a slightly yellow oil, 1-benzyloxycarbonyl-3,4-dibromo-3-ethylpiperidine, **8**,

(5.33 g, 98%). ¹H NMR: 1.14 (m, 3H); 2.02 (m, 3H); 2.78 (m, 1H); 3.33 (m, 2H); 4.19 (m, 2H); 4.62 (s, 1H); 5.17 (br s, 2H); 7.37 (m, 5H).

11

2-Carboxybenzyloxy-6-ethyl-2-azabicyclo[2.2.2]oct-5-ene-7,7-dicarboxylic acid diethyl ester, 11. Palladium on carbon (1.0 g, 10 % Pd) was suspended in reagent grade EtOAc (150 mL) and cooled to 0 °C on an ice bath with stirring. The mixture was then purged with hydrogen gas for 10 min. Diethyl ethoxymethylenemalonate (10.7 g, 49.4 mmol) was then added via syringe in one portion. The reaction mixture was allowed to warm to room temperature. After 2 h, TLC analysis indicated the the starting material had been consumed. The reaction mixture was purged with bubbling argon, then filtered through a pad of Celite, and rinsed with CH₂Cl₂. Removal of the solvent under vacuum afforded diethyl ethoxymethylmalonate as a clear, colorless oil (11.0g, 100%). ¹H NMR also indicated the presence of ca. 9% diethyl methylmalonate. The mixture was used as obtained in the following step.

Dibromo compound **8** (5.33 g, 13.2 mmol) and DABCO (5.13 g, 45.7 mmol) were dissolved in acetonitrile (100 mL). The solution was heated to reflux; a white precipitate formed rapidly. After ca. 2 h, TLC analysis showed that the starting material had been consumed. The mixture was cooled to room temperature (under argon) and the supernatant was decanted. The precipitate was rinsed (2x20 mL CH₂Cl₂) and the combined organic solutions were worked up between CH₂Cl₂ and water (2x20 mL CH₂Cl₂, brine), dried, and concentrated under vacuum to afford an orange, slightly cloudy oil (3.03)

g, 94% crude yield). Although the ¹H NMR spectrum of this oil was consistent with structure **9**, generally this material was not characterized but immediately used in the next step.

Crude dihydropyridine 9 and diethyl ethoxymethylmalonate were mixed and pump/purged with argon (3x). The slightly cloudy orange mixture was heated with stirring under argon under with a condenser at 100 °C overnight. The mixture was then cooled, and the volatiles were removed under vacuum. The residue was purified by flash column chromatography (hexane/ethylacetate 10:1-2:1) to afford 2-carboxybenzyloxy-6-ethyl-2-azabicyclo[2.2.2]oct-5-ene-7,7-dicarboxylic acid diethyl ester, 11, as a clear, slightly yellow oil (6.95g, >100%). The ¹H NMR of this material was consistent with the desired structure; the presence of polymeric ethyl esters was also revealed by the presense of broad multiplets near 1 and 4 ppm. The compound was used as obtained from the initial chromatography. ¹H NMR: the compound was found to exist as a mixture of rotamers that gave a complex spectrum; please see attached. Variable temperature NMR experiments supported the assignment as rotamers and not a mixture of regioisomers (Tc of dimethyl ester derivative in DMF- a^6 90 °C). IR (cm⁻¹): 858, 1021, 1108, 1149, 1245, 1412, 1704, 1735, 2879, 2980. HRMS: calc'd for C₂₃H₂₉NO₆: 415.1995; found: 415.2009.

5-ene-7,7-dicarboxylic acid, 12. The once-chromatographed diester **11** (6.95 g, ca. 13.2 mmol) was dissolved in reagent-grade ethanol (50 mL). An aqueous KOH solution (5 M, 50 mL) was added to give a homogenous yellow

solution which grew warm to the touch. This mixture was refluxed under argon with stirring for 2 h, until TLC analysis (4:1:1 toluene/formic acid/ethyl formate eluant) showed complete conversion of the starting material. The reaction mixture was cooled to room temperature and diluted with water (20 mL). The mixture was cooled to 0 °C on an ice bath, and acidified with 4 N HCl (ca. 90 mL) with stirring. The acidification was judged complete when the solution tested below pH 1 according to universal indicating paper, and a cloudy mixture was obtained. The mixture was extracted with ether (50 mL) and the layers separated. The aqueous layer was saturated with NaCl, and extracted with ether (5x30 mL). The combined extracts were dried (MgSO₄), filtered, and concentrated by rotary evaporation (bath temperature less than 30 °C) and exposure to vacuum to afford a viscous yellow oil. ¹H NMR indicated the presence of the desired diacid, along with appreciable quantities of ether and ethanol.

The crude diacid obtained above was dissolved in saturated aqueous NaHCO $_3$ (150 mL) to afford a cloudy white solution. Iodine (3.35 g, 13.2 mmol) was added and the mixture was stirred vigorously. The iodine slowly dissolved to afford a deep brown/purple solution. After stirring overnight, TLC (4:1:1 toluene/formic acid/ethyl formate eluant) showed complete consumption of the starting material. The mixture was carefully acidified with 4 N HCl (CAUTION! Vigorous bubbling!) until the pH was below 1 according to universal paper. The mixture was extracted with ether (100 mL). The aqueous layer was saturated with NaCl and extract with ether (5x30 mL). The combined organic layers were then decolorized with a small amount of saturated aqueous Na $_2$ S $_2$ O $_3$ solution, and washed with brine (10 mL). The solution was dried and concentrated by rotary evaporation; the last traces of solvent were removed by prolonged exposure to vacuum to afford iodolactone 12 as a pale yellow foam (4.26 g, 67% over the four steps from dibromide 8).

¹H NMR: the compound was found to exist as a mixture of rotamers that gave a complex spectrum; please see attached. IR (cm⁻¹): 734, 930, 964, 1078, 1118, 1176, 1308, 1346, 1422, 1709, 1794, 2944, 2974.

14

2-(cis-3'-hydroxyprop-1-enyl)phenylisothiocyanate, 14.2 (Note:

Failure to maintain the temperature of the initially obtained *cis* aldehyde at or below 0 °C prior to reduction results in the formation of appreciable amounts of the undesired trans isomer.) Quinoline (11.82 mL, 100 mmol) and BaCO3 (29.6 g, 150 mmol) were suspended in water (150 mL) and reagent-grade CH₂Cl₂ (150 mL) with rapid stirring at 0 °C in a ice bath. Thiophospene (7.62 mL, 100 mmol) was added dropwise over 15 minutes. After 35 min, TLC indicated complete consumption of the quinoline. The reaction mixture was filtered through a pad of Celite 545 with suction into an ice-cooled receiver, and the Celite was rinsed with small portions of ice-cold CH2Cl2. The mixture was partitioned between ice-cold brine and ice-cold CH2Cl2, and the layers were separated. The aqueous layer was washed (2x20 mL ice-cold CH2Cl2 and the combined deep-orange organic layers were diluted with ice-cold methanol (300 mL) and stirred rapidly at -15 °C in a methanol/ice bath. To this solution was added in small portions NaBH4 (1.89g, 50 mmol; CAUTION! Vigorous bubbling!) to afford a bright yellow solution. Immediately after adding the reductant, TLC indicated complete consumption of the aldehyde. The reaction was quenched with 1 M HCI (100 mL) and the layers separated. The organic layer was worked up between saturated NaHCO3 and CH2Cl2 (1x20ml CH₂Cl₂, brine) and dried. Concentration afforded a bad-smelling

²) Tokuyama, H; Yamashita, T.; Reding, M. T.; Kaburagi, Y.; Fukuyama, T. *J. Am. Chem. Soc.* **1999**, *121*, 3791.

deep orange oil, 19.27 g. This was recrystallized from hot CHCl₃/hexanes to afford one crop of fine needles (7.94 g) and the mother liquor was then concentrated and chromotographed (hexanes/EtOAc 10:1-1:2) to afford another portion of product (3.42 g). The title alcohol **14** was thus obtained as orange needles (11.36 g total, 60%) as a single *cis* diasteromer. ¹H NMR: 1.56 (s, 1H); 4.32 (br s, 2H); 6.08 (narrow quintet, 1H); 6.65 (t, *J*=11.6 Hz, 1H); 7.25 (m, 4H).

15

2-(*cis***-3'-hydroxyprop-1-enyl)aniline**, **15.**² Isothiocyanate **14** (3.83 g, 20 mmol) was dissolved in t-BuOH (75 mL). An aqueous KOH solution (5 M, 75 mL) was added. The mixture was stirred vigourously under argon at reflux for 2 h, until TLC indicated both the starting material and the presumed cyclothiocarbamate intermediate had been consumed. The reaction was cooled to room temperature, and worked up with saturated aqueous NH₄Cl solution (2x30 mL Et₂O, brine) and dried. Concentration afforded a sticky yellow solid, which was purified by flash column chromatography (hexanes/ethyl acetate 2:1-1:3 + 0.5% NEt₃) to afford the title aniline **15** as an off-white solid (2.12 g, 71%). ¹H NMR: 2.05 (s, 1H); 3.70 (br s, 2H); 4.23 (d, \mathcal{L} 6.8 Hz, 2H); 6.01 (m, 1 H); 6.47 (d, \mathcal{L} 5.11.6 Hz, 1 H); 6.95 (d, \mathcal{L} 5.8.0 Hz, 1H); 6.75 (m, 2H); 7.10 (t, \mathcal{L} 6.8 Hz, 1H).

2-alkenyl anilide 16. lodolactone acid 12 (3.9 g, 8.04 mmol), aniline 15 (1.00 g, 6.74 mmol), and water-soluble carbodiimide (1.7 g, 8.75 mmol) were dissolved in CH₂Cl₂ (25 mL) under argon. NEt₃ (1.4 mL, 10.11 mmol) was added via syringe and the mixture was stirred at room temperature until TLC indicated complete consumption of the aniline (1.5 h). The reaction mixture was partitioned between 1 N HCl and CH₂Cl₂ (50 mL each); the layers were separated and the organic layers were washed (1x 50 ML each 1 N HCL, sat'd NaHCO₃, brine), dried, and concentrated to afford the desired crude anilide as a white foam (4.55 g). The crude material was immediately dissolved in a mixture of Ac₂O (5 mL, 53 mmol) and pyridine (5 mL, 62 mmol) and stirred for 2 h at room temperature. The mixture was then partitioned between ether and 1 N HCL; the layers were separated and the organic layers were washed (2 x 20 mL sat'd NaHCO₃, 1 x 20 mL brine), dried, and concentrated to afford a viscous orange-brown oil (4.95 g), which was purified by flash column chromatography (10:1-2:1 hexanes/EtOAc + 0.5% NEt₃) to afford the anilide 16 as a colorless foam (3.29 g, 74 %). ¹H and ¹³C NMR: the compound was found to exist as a mixture of rotamers that gave complex spectra; please see attached. IR (cm⁻¹): 764, 963, 1234, 1305, 1420, 1535, 1708, 1737, 1764, 2943, 2975, 3325. HRMS: calc'd for C₃₀H₃₁IN₂O₇: 658.1176; found: 658.1197.

17

De-iodolactonized methyl ester anilide 17. Anilide 16 (3.29 g, 5.00 mmol) was dissolved in CH₂Cl₂ (25 mL) under argon. Powdered zinc (3.3 g, 50 mmol) was added, followed by glacial acetic acid (5.7 mL, 100 mmol) with rapid stirring. The reaction mixture warmed slightly, and was stirred at room temperature until TLC indicated complete consumption of the starting material (30 min). The reaction mixture was passed through Celite to remove excess zinc; the Celite pad was rinsed with several portions each of CH₂Cl₂ and ether. The filtrate was washed (3 x 20 mL 1 N HCl), dried, and filtered. The solution volume was reduced to ca. 100 mL. To this gently swirled solution was added an ethereal solution of diazomethane (generated at 0 °C from Nmethyl-N-nitrosourea using 50% aqueous NaOH). The diazomethane solution was added at room temperature until the bright yellow color persisted for 2-3 min. The yellow solution was quenched with a small amount of AcOH, and then stirred with sat'd NaHCO3. The layers were separated and the organic layer was washed (brine), dried, and concentrated to afford a bright yellow oil (3.02 g). This was purified by flash column chromatography (10:1-2:1 hexanes/EtOAc + 0.5% NEt₃) to afford the desired methyl ester 17 as a clear, viscous, nearly colorless oil (2.26 g, 83%). ¹H and ¹³C NMR: the compound was found to exist as a mixture of rotamers that gave complex spectra; please see attached. IR (cm⁻¹): 760, 1030, 1250, 1451, 1526, 1581, 1702, 1740, 2962, 3382. HRMS: calc'd for C₃₁H₃₄N₂O₇: 546.2366; found: 546.2363.

Thioanilide 18. Anilide 17 (2.26 g, 4.13 mmol) was dissolved in toluene (50 mL) under argon. Lawesson's reagent (5.02 g, 12.4 mmol) was added, followed by pyridine (0.17 mL, 2.1 mmol). The mixture was stirred at reflux under a condenser and argon balloon overnight, after which time TLC showed complete consumption of the starting material. The mixture was cooled to room temperature, and worked up (sat'd NaHCO₃/ether, brine), dried, and concentrated. The resulting bad-smelling bright-yellow oil was purified by flash column chromatography (5:1-1:5 hexanes/CH₂Cl₂ + 1% MeOH) to afford thioanilide 18 as a nearly odorless, pale-yellow foam (2.00 g, 86%). ¹H and ¹³C NMR: the compound was found to exist as a mixture of rotamers that gave complex spectra; please see attached. IR (cm⁻¹): 761, 1101, 1238, 1370, 1417, 1507, 1699, 1739, 2962. HRMS: calc'd for C₃₁H₃₄N₂O₆S: 562.2138; found: 562.2125.

Indole 19. Thioanilide **18** (414 mg, 0.74 mmol) was dissolved in 1-propanol (5 mL) under argon along with AIBN (133 mg, 0.81 mmol), hypophophorous acid (30% aqueous solution, 1.4 mL, 7.4 mmol), and NEt₃ (1.55 mL, 11.1 mmol). The mixture warmed slightly, and was stirred in a 90 °C oil bath under a reflux condenser and argon balloon. After 45 min, TLC indicated complete consumption of the starting material. The mixture was cooled to RT and

diluted with ether (20 mL). Workup (1 N HCl, 1 N NaOH, brine), drying, and concentration afforded a yellow oil, which was purified by flash column chromatography (10:1-5:1 hexanes/EtOAc + 0.5% NEt₃) to give the desired indole **19** as a clear, faintly yellow oil (198 mg, 50%). ¹H NMR: 1.06 (t, *J*=7.6 Hz, 3H); 1.91 (d, *J*=10.8 Hz, 1H); 2.04 (s, 3H); 2.18 (m, 2H); 2.9-3.2 (m, 5H); 3.54 (s, 3H); 3.60 (m, 1H); 4.14 (m, 1H); 4.25 (m, 1H); 5.12 (q, *J*=15.6, 10.8 Hz, 2H); 5.48 (s, 1H); 6.11 (d, *J*=5.6 Hz, 1 H); 7.09 (t, *J*=8.0 Hz, 1H); 7.17 (t, *J*=7.2 Hz, 1H); 7.38 (d, *J*=8.0 Hz, 1H); 7.60 (d, *J*=8.0 Hz, 1H); 9.96 (s, 1H). ¹³C NMR: 11.4, 21.1, 23.4, 24.3, 25.1, 26.5, 30.8, 34.9, 48.4, 52.6, 53.5, 63.8, 67.6, 108.4, 111.6, 118.5, 119.3, 121.8, 126.7, 127.7, 128.1, 128.5, 128.7, 135.2, 135.5, 136.1, 144.5, 156.9, 171.0, 171.4, (1 obscured peak). IR (cm⁻¹): 737, 1083, 1244, 1419, 1457, 1672, 1698, 1738, 2962, 3298. HRMS: calc'd for C₃₁H₃₄N₂O₆: 530.2417; found: 530.2406.

Mesylate 20. Acetate-indole 19 (166 mg, 0.29 mmol) was dissolved in dry methanol (5 mL) under argon. K_2CO_3 (powdered anhydrous, 200 mg, 1.45 mmol) was added and the mixture was stirred vigorously for 30 min at room temperature, at which time TLC analysis showed complete consumption of the starting material. The mixture was partitioned between ether and water, and worked up (ether, brine), dried, and concentrated top afford a white foam (152 mg). The crude material was immediately dissolved in dry CH_2Cl_2 (2 mL) and treated with mesyl chloride (27 μL, 0.35 mmol) and NEt_3 (53 μL, 0.38 mmol). The mixture was stirred for 20 minutes at room temperature, at which time TLC analysis indicated complete consumption of the free primary alcohol. The

mixture was partitioned between 1 N HCl and ether, and worked up (ether, NaHCO₃, brine), dried, and concentrated to afford the crude product as a yellow oil (176 mg). This material was purified by flash column chromatography (10:1-2:1 hexanes/EtOAc + 0.5% NEt₃) to afford the desired mesylate **20** as a white foam (135 mg, 82%). ¹H NMR: 1.06 (t, *J*=6.8 Hz, 3H); 1.88 (d, *J*=12.8 Hz, 1H); 2.17 (m, 2H); 2.78 (s, 3H); 2.93 (m, 2H); 3.21 (m, 5H); 3.57 (s, 3H); 4.25 (m, 1H); 4.35 (m, 1H); 5.17 (s, 2H); 5.62 (s, 1H); 6.12 (br d, *J*=6.0 Hz, 1H); 7.11 (t, *J*=7.6 Hz, 1H); 7.21 (t, *J*=7.6 Hz, 1H); 7.38 (d, *J*=7.6 Hz, 1H); 7.52 (d, *J*=7.6 Hz, 1H); 10.06 (s, 1H). ¹³C NMR: 11.3, 24.9, 26.5, 30.7, 34.9, 37.2, 48.4, 52.7, 53.4, 53.5, 67.6, 68.9, 106.6, 111.8, 118.0, 119.6, 122.0, 126.8, 127.7, 128.1, 128.3, 128.4, 135.1, 136.0, 136.1, 144.4, 157.0, 171.5 (2 obscured peaks). IR (cm⁻¹): 737, 952, 1174, 1355, 1421, 1458, 1671, 1735, 2962, 3296.. HRMS: calc'd for C₃₀H₃₄N₂O₆S: 566.2087; found: 566.2088.

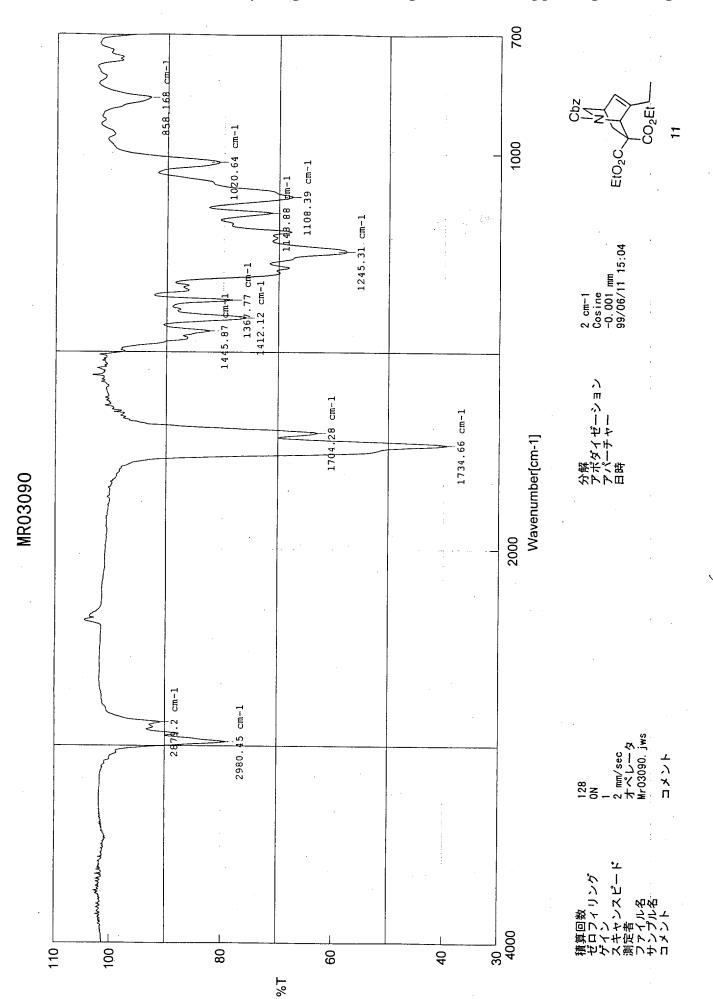
(±)-Catharanthine, 1.3 Mesylate 20 (82 mg, 0.14 mmol) was dissolved in EtOAc (passed through Al₂O₃, 0.5 mL) along with Pd(OAc)₂ (11 mg, 0.05 mmol) under argon. Dry ethanol (0.5 mL) was added and the mixture was stirred at room temperature for 3 min (orange solution slowly turned red/brown). Triethylsilane (50 μL, 0.29 mmol) was added via syringe, and the mixture bubbled, warmed, and rapidly turned black. NEt₃ (20 μL, 0.14 mmol) was added immediately. The mixture was stirred at room temperature for 15 min, at which time TLC analysis indicated complete consumption of the starting material. The reaction mixture was diluted with CH₂Cl₂, and filtered through Celite to remove the precipitated palladium. The filtrate was worked

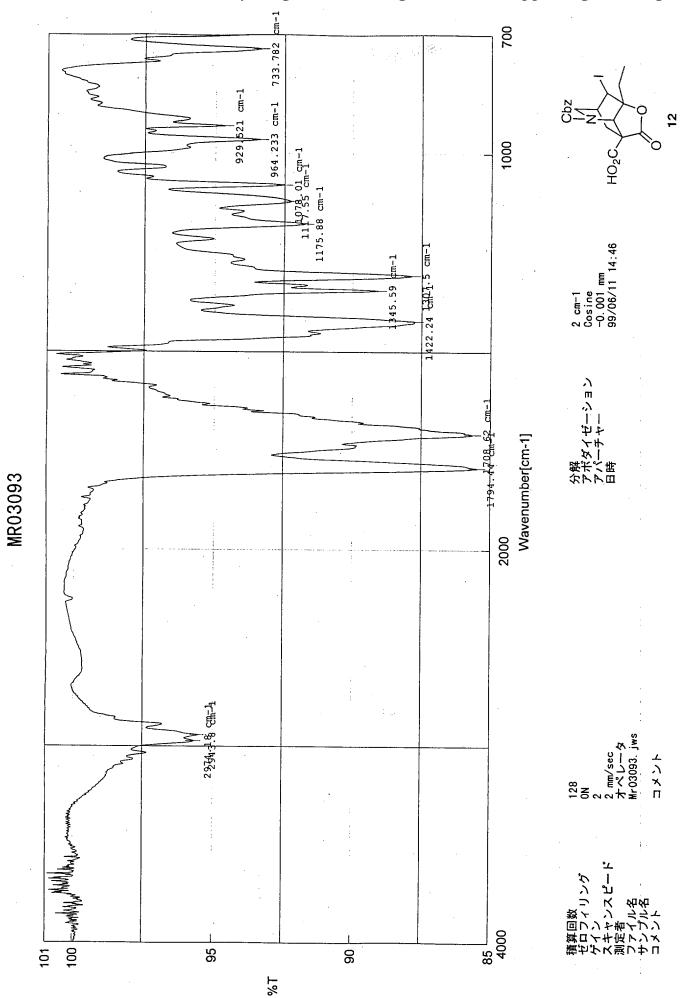
³Kuehne, M. E.; Bornmann, W. G.; Earley, W. G.; Marko, I. J. Org. Chem. 1986, 51, 2913.

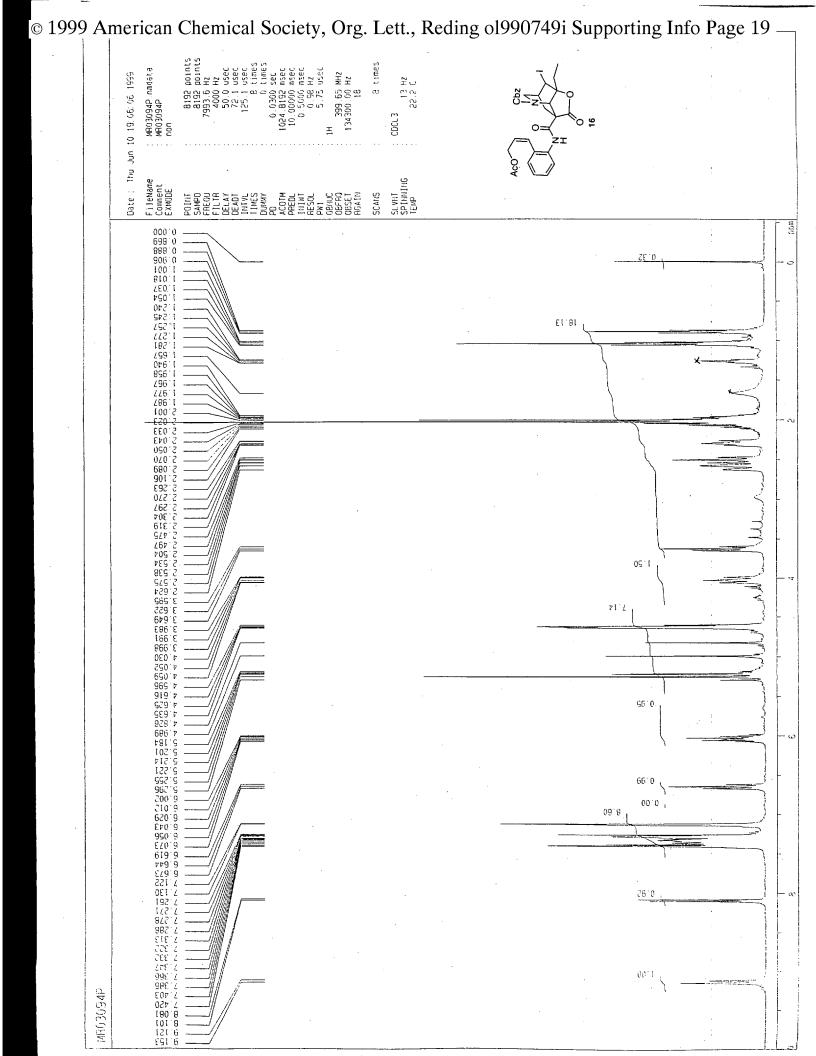
up (CH₂Cl₂, NaHCO₃), dried, and concentrated. The crude material was purified by pTLC (two 200 x 65 x 0.5 mm plates, double elution with 1:1 hexanes/EtOAc + 0.5% NEt₃) to afford (\pm)-catharanthine, **1**, as a white powder (45 mg, 96%). ¹H NMR: 1.07 (t, J=6.8 Hz, 3H); 1.77 (dd, J=13.6, 3.6 Hz, 1H); 2.09 (m, 1H); 2.31 (m, 1H); 2.72 (br d, J=6.8 Hz, 1H); 2.81-2.94 (m, 3H); 3.25-3.40 (m, 3H); 3.56 (m, 1H); 3.73 (s, 3H); 4.17 (s, 1H); 5.92 (dm, 1H); 7.08-7.16 (overlapping triplets, 2H); 7.23 (d, J=7.6 Hz, 1H); 7.48 (d, J=7.6 Hz, 1H); 7.68 (s, 1H). ¹³C NMR: 10.6, 21.4, 28.2, 30.7, 38.7, 49.3, 52.3, 53.0, 55.4, 61.9, 110.4, 110.7, 118.2, 119.4, 121.8, 123.5, 129.0, 134.9, 136.4, 149.4, 174.2. IR (cm⁻¹): 741, 909, 1078, 1267, 1460, 1713, 2844, 2961, 3375. HRMS: calc'd for C₂₁H₂₄N₂O₂: 336.1838; found: 336.1846. The structure was further confirmed by ¹H-¹H COSY, DEPT, and HMQC (¹H-¹³C correlation) NMR measurements (data not shown).

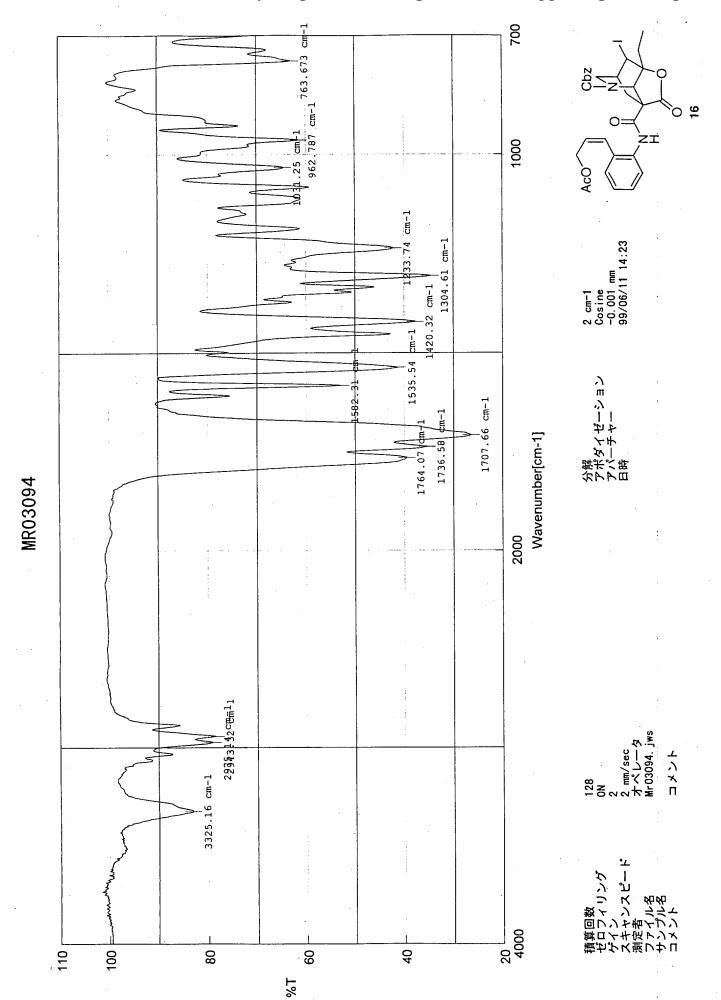
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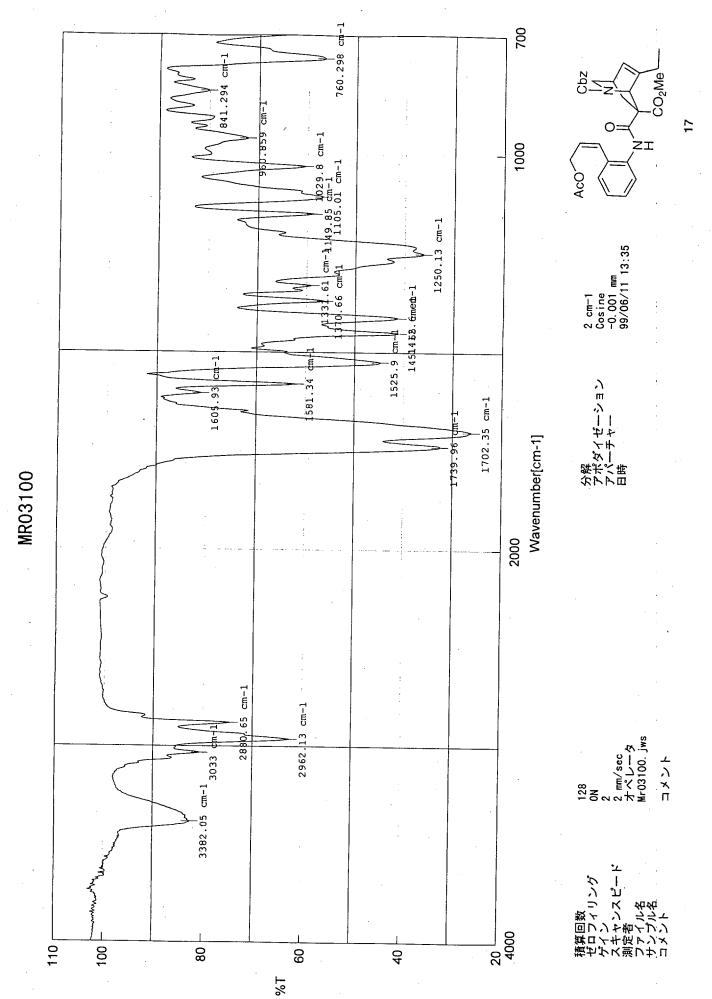
The ¹H NMR, ¹³C NMR, and IR spectra for selected compounds are reproduced on the following pages.

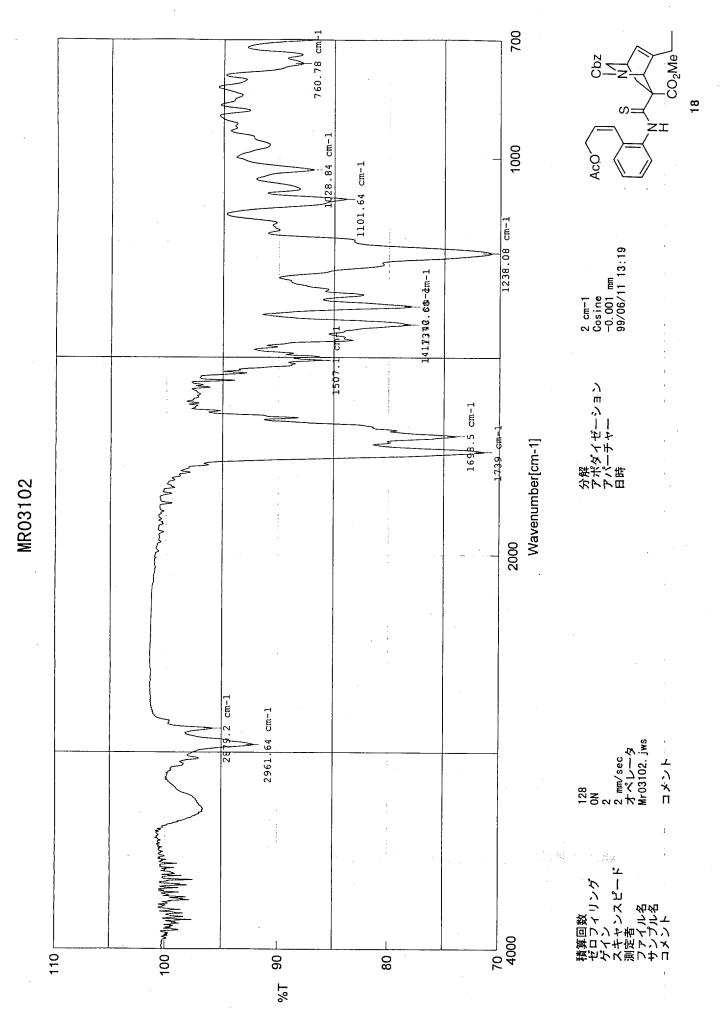


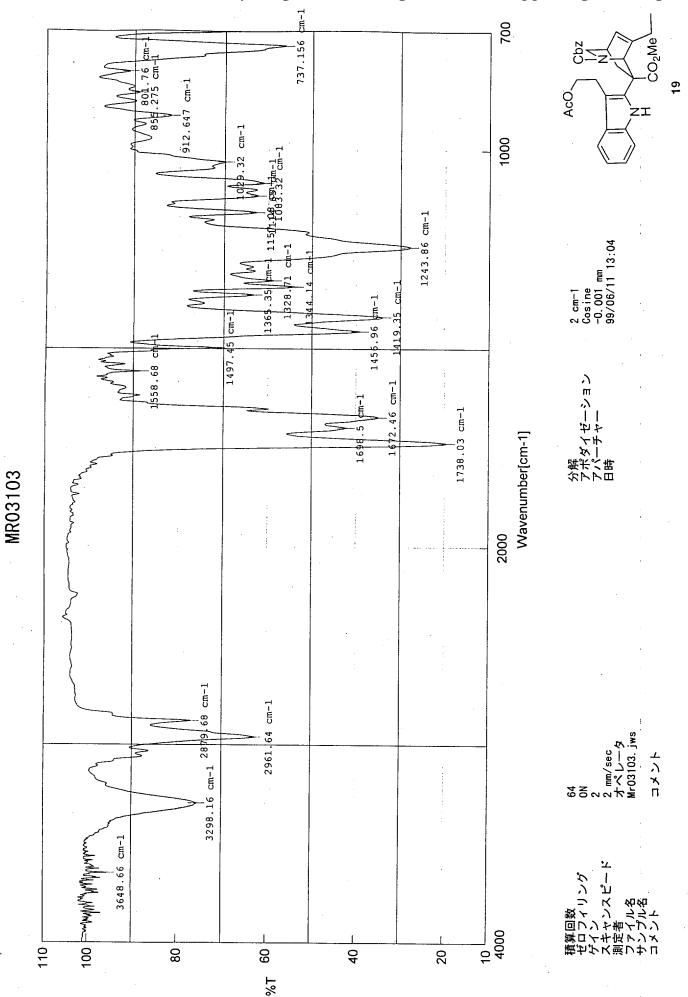


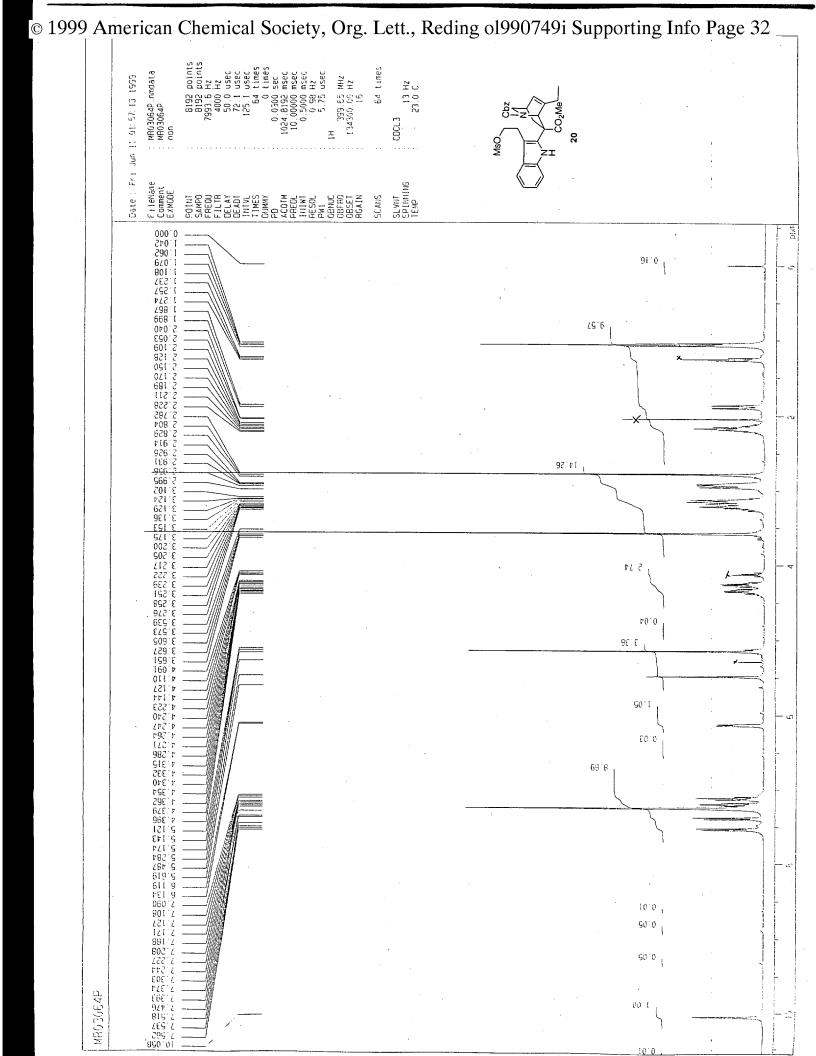












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